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Solution From Coulomb's law of electrostatic attraction we know that the PE between two charges Q_1 and Q_2 separated by a distance r is given by $PE = \frac{1}{4\pi\epsilon_0} \frac{Q_1 Q_2}{r}$ First we consider the interaction between Na^+ ion and $6Cl^-$ ions at distance r Applying Coulomb's law we have

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$Q_1 Q_2 = (6e)(+e) = 6e^2$ PE1 = = = $4\pi \epsilon_0 r$ or $4\pi \epsilon_0 r$
 $4\pi \epsilon_0 r$ Similarly, we now consider 12 Na+ ions as second nearest neighbors at a distance r
 $Q_1 Q_2 = (+12e)(+e) = 12e^2$ PE2 = = $4\pi \epsilon_0 r$ or $4\pi \epsilon_0 r$ and Cl ...

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metal (BCC) with a density of 0.9712 g cm^{-3} .
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