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Chemistry 12 - Chapter 7 Quiz.

True/False. Indicate whether the
sentence or statement is true or false.

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T F. 1. Chemical equilibrium means that all chemical reactions have stopped. T F. 2. Equilibrium can only occur in a closed system.

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Equilibrium 7.1-4. Section 7.1
Questions, page 428 1. (a) A sealed

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bottle of pop is a dynamic equilibrium because the rate of conversion between carbon dioxide and carbonic acid is equal in both directions in the closed system. (b) When I open the pop bottle, carbon dioxide leaves the bottle, so the equilibrium conditions no longer exist and a new equilibrium

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must form between the pop and the atmosphere.

~~Section 7.1: Equilibrium Systems Mini
Investigation: The ...~~

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7.2-4 3. Given: $4 \text{ Fe(s)} + 3 \text{ O}_2\text{(g)}$ #!!!

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$2 \text{Fe}_2\text{O}_3(\text{s})$; volume = 2.0 L;
quantities at equilibrium: $\text{Fe}(\text{s}) = 1.0$
mol; $\text{O}_2(\text{g}) = 1.0 \times 10^{-3}$ mol; $\text{Fe}_2\text{O}_3(\text{s}) = 2.0$ mol
Required: K
Solution:
Step 1. Write the equilibrium law
equation using the balanced chemical
equation. $K = \frac{1}{[\text{O}_2(\text{g})]^3}$
Step 2.

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~~Section 7.2: Equilibrium Law and the
Equilibrium Constant ...~~

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Section 7.5: Quantitative Changes in
Equilibrium Systems The

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Organic Chemistry ... 2.7 Fats and Oils

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Careers in Chemistry Chapter 2 Lab
Activities Investigation 2.1.1:
Identification of Plastics Activity 2.1.2:
Making Guar Gum Slime ... Chapter 7
Review Chapter 8: Acid–Base
Equilibrium

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Gr 12 U1- Organic Chemistry; Grade
12 U2- Structure and Properties of
Matter; ... Below are all of the
resources for chapter 7 and 8. This is
an important unit because there are a
lot of questions on the exam and there
are a lot of labs in this unit. ... 7.1 p.
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and the ...

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Chapter 1, Organic Compounds, introduces the historical concept of organic compounds as those compounds produced by living things. Students then learn that the current concept of organic chemistry is the study of carbon compounds. Each section of the chapter addresses a

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different group of organic compounds,
related by structure.

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Publisher McGraw-Hill Publishing
Company ISBN 978-0-07802-152-7

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Ltd. Chapter 7: Electric Fields 7.6-2

Determine the excess of electrons: $N = \frac{q}{e} = \frac{1.589 \times 10^{-18} \text{ C}}{1.602 \times 10^{-19} \text{ C}}$

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~~Revision Answers~~
N=10 Statement: The charge on the oil drop is -1.6×10^{-18} C. The oil drop has an excess of 10 electrons, or $-10e$. 3. Given: $? = 1.0 \times 10^2$ N/C; $m = 2.4 \times 10^{-15}$ kg; $e = 1.602 \times 10^{-19}$ C; $g = 9.8$ m/s² Required: q

~~Section 7.6: The Millikan Oil Drop~~

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Chapter 9 Nelson Solutions
Manual.pdf. Page 1. Page 2. Page 3.
Page 4. Page 5. Page 6. Page 7. ... A
copy of the physical constants is
provided on page 26. The Chemistry
7–12 test is designed to include a total
of 100 multiple-choice. 259.

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mL ...

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The values can now be substituted

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Review Questions
into the equilibrium equation for the
ionization of a weak acid: $K_a = \frac{[H^+][A^-]}{[HA]}$

~~Section 8.7: Acid-Base Titration
Tutorial 1 Practice, page 547~~

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