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CHAPTER 2 REVIEW Measurements
and Calculations SECTION 1 SHORT
ANSWER Answer the following
questions in the space provided. 1.

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Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model. observation and data a. A research team records the rainfall in inches

2 Measurements and Calculations

Modern Chemistry 105 Chapter Test

Name Class Date Chapter Test A,

continued Use this figure to answer

questions 7 and 8. _____ 7. A solution

containing 35 g of Li_2SO_4 dissolved in 100 g of water is heated from 10°C

to 90°C . According to information in

the figure, this temperature change

would result in a. an additional 5 g of Li_2SO_4 in ...

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20 shows that 2 molecules (made of 4

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atoms) of hydrogen and 1 molecule (made of 2 atoms) of oxygen produce 2 molecules of water. The total mass of the product, water, is equal to the sum of the masses of each of the reactants, hydrogen and oxygen. What parts of Dalton's atomic theory are illustrated by this reaction?

3 Atoms: The Building Blocks of Matter

Chemistry (4th Edition) Burdge, Julia
Publisher McGraw-Hill Publishing
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PROBLEMS Write the answers to the questions on the line to the left, and show your work in the space provided.

25. 26. MODERN CHEMISTRY The atomic number of nickel-60 is 28. How many neutrons does this isotope have? Carbon-14 has 8 neutrons.

What is the atomic number of carbon-14? 11 CHAPTER 3 TEST

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Change MIXED REVIEW SHORT

ANSWER Answer the following
questions in the space provided. 1.

Classify each of the following as a
homogeneous or heterogeneous
substance. homogeneous a. sugar
homogeneous d. plastic wrap

homogeneous b. iron filings

heterogeneous e. cement sidewalk

heterogeneous c. granola bar 2. For
each type of investigation, select the
most appropriate ...

1 Matter and Change - HUBBARD'S CHEMISTRY

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER

Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

10 States of Matter - Ms. Agostine's

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Algebra 1: Common Core (15th Edition) Charles, Randall I. Publisher Prentice Hall ISBN 978-0-13328-114-9

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Chemistry 179 Organic Chemistry

CHAPTER 22 REVIEW Organic

Chemistry SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1.

Name two types of carbon-containing molecules that are not organic. _____

2. _____ Carbon atoms form bonds readily with atoms of ...

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