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Copyright © 2004 by College Entrance Examination Board. All rights reserved. Visit apcentral.collegeboard.com (for AP professionals) and www.collegeboard.com/apstudents (for AP students and parents). 4 Question 2 2 Fe(s) + 3 2 O2(g) → Fe2O3(s) ΔHfD = -824 kJ mol-1

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CH₃CH₂COOH(aq) + H₂O(l) ⇌ CH₃CH₂COO⁻(aq) + H⁺(aq)
Propanoic acid, CH₃CH₂COOH, is a carboxylic acid that reacts with water according to the equation above. At 25 °C the pH of a 50.0 mL sample of 0.20 M CH₃CH₂COOH is 2.95.

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at 700. Question 1 Sample: 1A Score: 10 This response earns a perfect score of 10 points: 2 points for part (a), 1 point for part (b) (i), 2 points for part (b) (ii), 2...

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Question 8 Answer the following questions about carbon monoxide, $\text{CO}(\text{g})$, and carbon dioxide, $\text{CO}_2(\text{g})$. Assume that both gases exhibit ideal behavior. (a) Draw the complete Lewis structure (electron-dot diagram) for the CO molecule and for the CO_2 molecule. 1

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point for each correct, complete Lewis structure

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